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Question 1 . 1. Each of
three beakers contains

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25.0 mL of a 0.100 M
solution of HCl, NH₃,
or NH₄Cl, as shown
above.

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each. Your responses to
these questions will be
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student is assigned the task of determining the mass percent of silver in an alloy of copper and silver by dissolving a

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sample of the alloy in excess nitric acid and then precipitating the silver as AgCl. First the student prepares 50. mL of 6 M HNO₃ .

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energy to overcome the
stronger hydrazine,
resulting m a higher
boiling pomt- 111 based
on N₂H₄ IS a polar
tnolecule with London

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Intermolecular forces, dipole—dipole forces, and hydrogen bonding between molecules, whereas C_2H_6 is nonpolar and only has London dispersion forces between molecules. It is important to the normal boiling point of N_2H_4 is $114.0\text{ }^\circ\text{C}$, whereas the normal boiling point of C_2H_6 is $-88.6\text{ }^\circ\text{C}$. Explain, in terms of the ...

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worth 10 points, and 4
short questions, each
worth 4 points.

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mol sample of
methanol, CH_3OH , is
placed in a 15.0 L
evacuated rigid tank and
heated to 327°C .

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RESPONSE
QUESTIONS**

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The free-response section of the AP Chemistry exam seems intimidating because it's longer than free-response sections on most other AP tests, and it includes lots of calculations and experiments that you have to interpret. The most important thing you can do is remain calm and stay focused

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and methodical in your
approach to each
question.

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AP Chemistry FRQ: How to Ace the Free Response Questions

The AP Chemistry
Exam covers a full-year
introductory college
course in chemistry with
laboratory. The multiple-
choice questions in
Section I cover the

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breadth of the
curriculum. Section II,
the free-response part of
the exam, includes three
quantitative problems,
one question on writing
chemical reactions and
predicting products, and
two essays.

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from the 2011 AP
Chemistry exam. The
question deals with
Lewis dot-diagrams,
Intermolecular Forces,
reactions with water...

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6. In an experiment, all the air in a rigid 2.0 L flask is pumped out. Then some liquid ethanol is injected into the sealed flask, which is held at 35 ° C.

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