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9 Study Guide

Answers

Chapter 9

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Answers

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Answers

Chapter 1:

Matter and

Change (Chem in

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Matter and

Change **Matter**

and Change part

1 (Chem 1 H)

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Chemistry

GCSE Science

Revision

Chemistry \ "The

Three States of

Matter \ "The

Creation of

Chemistry - The

Fundamental

Laws: Crash

Course Chemistry

#3 Matter and

Change Chapter 1

— Introduction:—

Matter and

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Chemistry

~~Matter And~~

States of matter

| States of

matter and

intermolecular

forces |

Chemistry | Khan

Academy Grade 9

Chemistry Lesson

1 - Matter and

the Particle

Theory ~~What Is~~

Matter? ~~The~~

Dr. Binoes Show

Page 7/46

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Chemistry

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~~Videos For Kids~~

~~Peekaboo Kidz~~

~~Chemistry Matter~~

~~and Change Part~~

~~1 Particulate~~

~~nature of matter~~

~~- Part-1 -~~

~~Chemistry~~

Changes of

States of Matter

Chemical and

Physical Changes

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Chemistry

Matter And

• Solid Liquid

Gas 10 Amazing

Experiments with

Water

Pure Substances

and Mixtures!

(Classification

of Matter)

Physical and

Chemical Changes

Understanding

Atoms, elements,

Page 9/46

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Chemistry

and molecules

Part #1 (9min)

Basic Chemistry

Series - 1.2

What is matter?

- HD Properties

of Matter

Physical and

Chemical

Changes:

Chemistry for

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Changes States

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Chemistry

~~of Matter And~~

~~Solids, Liquids,~~

~~Gases \u0026~~

~~Plasma~~

~~Chemistry Change~~

~~of State |~~

~~Matter | Physics~~

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States of Matter

and Changes of

State - Science

for Kids GCSE

Chemistry -

States of Matter

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Matter And Changing

State #20 Pure
Substances and
Mixtures,

Elements \u0026

Compounds,

Classification

of Matter,

Chemistry

Examples,

Chemistry/ICSE/C

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Classifying

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~~Matter With~~

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1 - Introduction

to Chemistry 1.1

The Stories of

Two Chemicals

Ozone Layer,

atmosphere,

ozone formation,

chlorofluorocarb

ons, CFC's 1.2

Chemistry and

Matter Chemistry

Central Science

Chapter

Page 15/46

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Chemistry

Summaries -

Chemistry Matter
and Change

Chapter 1

Section 2 MATTER

and Its

Properties

- Anything that

has MASS and

VOLUME. ATOMS

and MOLECULES

the basic

building blocks

of matter.

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Chemistry

ATOM— the smallest unit of an element that still maintains the properties that element.

ELEMENT— a pure substance made up of only one type of atom.

**MATTER and
CHANGE Chapter 1
Section 1**

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Chemistry

Matter and

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6

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Periodic Trends

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Matter and Changes

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3 35 Section 3.1

Properties of

Matter pages

70–75 Problem-

Solving Lab 1.

Explain why the

flow of a

compressed gas

must be

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controlled for
practical and
safe use.

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Effect, Movie

Ozone

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6 - The Periodic

Table and

Periodic Law

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Matter And Section 6.1

Development of Change Chapter 9 Study Guide the Modern Periodic Table

- Answers
1. octaves
 2. eight
 3. nine
 4. accepted
 5. Dmitri Mendeleev
 6. atomic mass
 7. elements
 8. atomic number
 9. Henry Moseley
 10. protons
 11. periodic law
 - 12.

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properties 13.

14 . . .

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answers

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Assessment

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Matter and

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5 25 Electrons

in Atoms

Reviewing

Page 27/46

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Chemistry

Vocabulary Match
the definition
in Column A with
the term in
Column B. Column
A Column B 1.

The set of
frequencies of
the
electromagnetic
waves emitted by
the atoms of an
element 2. The
minimum amount

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combines the

elements

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incorporates
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Matter - 2.1

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Chapter 2:

Matter and

Change.

extensive

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intensive

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Chemistry

Matter And

extensive

properties

(examples)

intensive

property

(examples) a

property that

depends on the

amount of matter

in a sample. a

property that

depends on the

type of matter

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in a sample
not... mass,
volume, length,
weight.

Answers

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and change

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moles of known

substance C H C

H C H O O O CO

CO CO H O H O H

O number of

moles of unknown

substance O CO H

O C H CO H O C H

O H O C H O CO

use mole ratio 5

mol O 2 1 mol C

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Chemistry

3 H 8 3 mol CO 2

1 mol C 3 H 8 4

mol H 2 O 1 mol

C 3 H 8 1 mol C

3 H 8 5 mol O 2

MATH SKILLS

TRANSPARENCY

MASTER 16

A "chemical change" is a change that produces matter with a different

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composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of

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Compounds Quite
different
properties than
their component
elements.

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