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SHORT ANSWER Answer the
following questions in the space
provided. 1. Explain the difference
between the mass number and the
atomic number of a nuclide. Mass

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number is the total number of protons and neutrons in the nucleus of an isotope. Atomic number is the total number of protons in the nucleus of each atom of an element. 2.

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matter, changes that occur in matter, and the relationships between matter and energy analytical chemistry—the identification of the composition of materials biochemistry—the study of the chemistry of living things

1 Matter and Change - HUBBARD'S

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CHEMISTRY Answers

Title: Study GuideChapter 5-21

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Study GuideChapter 5-21 Answer Key
SHORT ANSWER Answer the
following questions in the space

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provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

observation and data a. A research team records the rainfall in inches per day in a prescribed area of the rain forest. The

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2 Measurements and Calculations
phase energy (E_{ph}) -energy stored in
system due to arrangement of
particles that exert attractions on one
another. (energy account involved
when phase changes occur)
-Attractions result in decrease of

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energy of a system of particles. -As particles become more tightly bound, their phase energy is lowered. -Solids possess lowest phase energy, and liquids possess more because the particles move freer.

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SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an

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element that can exist either alone or in molecules containing the same or different elements is the .atom 3.

3 Atoms: The Building Blocks of Matter
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Answers Chapter 2

the intermolecular attractions resulting from the constant motion of electrons and the creation of instantaneous dipoles; forces act between all atoms AND molecules; forces are NOT present in intramolecular forces; increases with atomic mass or molar

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